

PHOTOISOMERIZATION OF PHTHALOYL SYSTEM TO ALKYLIDENE PHTHALIDES

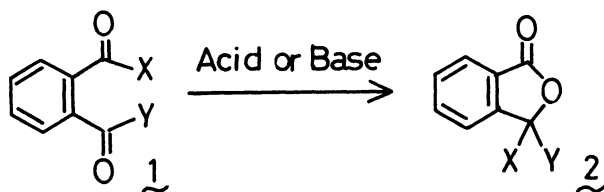
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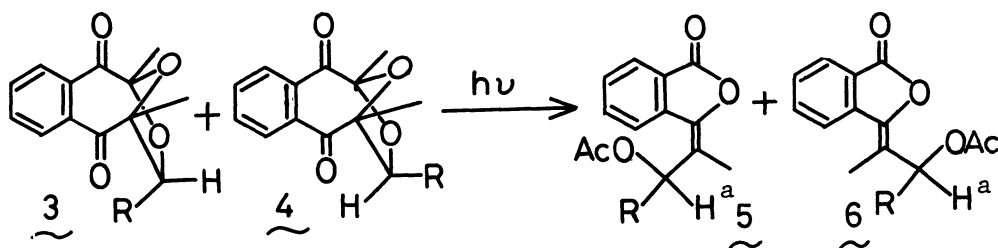
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Irradiation of a benzene solution of bicyclic diketones 3 and/or 4 gave alkylidene phthalides 5 and 6 in good yields. Structural effects on the photoisomerization were discussed.

Acid- or base-catalyzed isomerization of phthaloyl system 1 to *as*-phthalyl system 2 has been long known¹⁾, but a photochemical counterpart has not been reported so far. In the course of our studies on the photochemical reactions of epoxynaphthoquinones, we found that bicyclic diketones 3a-h and 4a-h, upon irradiation, readily underwent photoisomerization, giving alkylidene phthalides 5a-h and 6a-h in good yields. The present work is concerned with investigation on the photoisomerization of a series of phthaloyl compounds 3a-h and 4a-h.



Irradiation of a benzene solution (0.02M) of 3a-h and/or 4a-h²⁾ with a high-pressure Hg lamp for 30 hours gave a mixture of alkylidene phthalides 5a-h and 6a-h. These phthalides were separated by chromatography on silica gel and their structural assignments



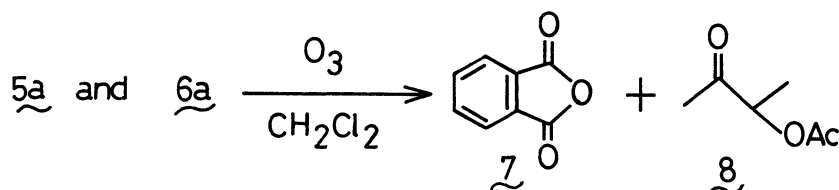
- a, R=Me
- b, R=Et
- c, R=Pr
- d, R=i-Pr
- e, R=p-MeO-C₆H₄
- f, R=p-Me-C₆H₄
- g, R=Ph
- h, R=p-Cl-C₆H₄

Table I. Photoisomerization of 3 and/or 4 to 5 and 6.

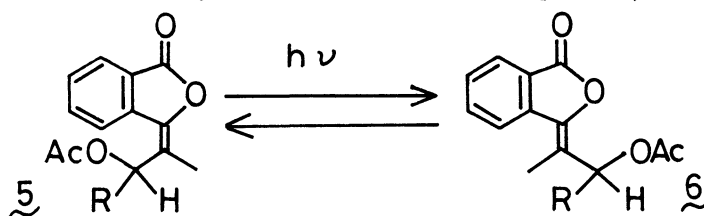
Reactant	Total Yield (%) ^{a)}	<u>5/6</u> ^{b)}	Products (mp, °C)
<u>3a</u> + <u>4a</u>	80	1.1	<u>5a</u> (84.5) <u>6a</u> (88)
<u>3b</u> + <u>4b</u>	75	1.2	<u>5b</u> (88) <u>6b</u> (90)
<u>3c</u> + <u>4c</u>	73	1.3	<u>5c</u> (105) <u>6c</u> (oil)
<u>3d</u> + <u>4d</u>	66	1.0	<u>5d</u> (oil) <u>6d</u> (126)
<u>3e</u> + <u>4e</u>	92	1.2	<u>5e</u> (119) <u>6e</u> (124)
<u>3f</u> + <u>4f</u>	90	0.9	<u>5f</u> (127) <u>6f</u> (144)
<u>3g</u> + <u>4g</u>	86	0.9	<u>5g</u> (135) <u>6g</u> (130)
<u>3h</u> + <u>4h</u>	86	1.0	<u>5h</u> (153) <u>6h</u> (148)

a), Isolated yields. b), Determined by HPLC.

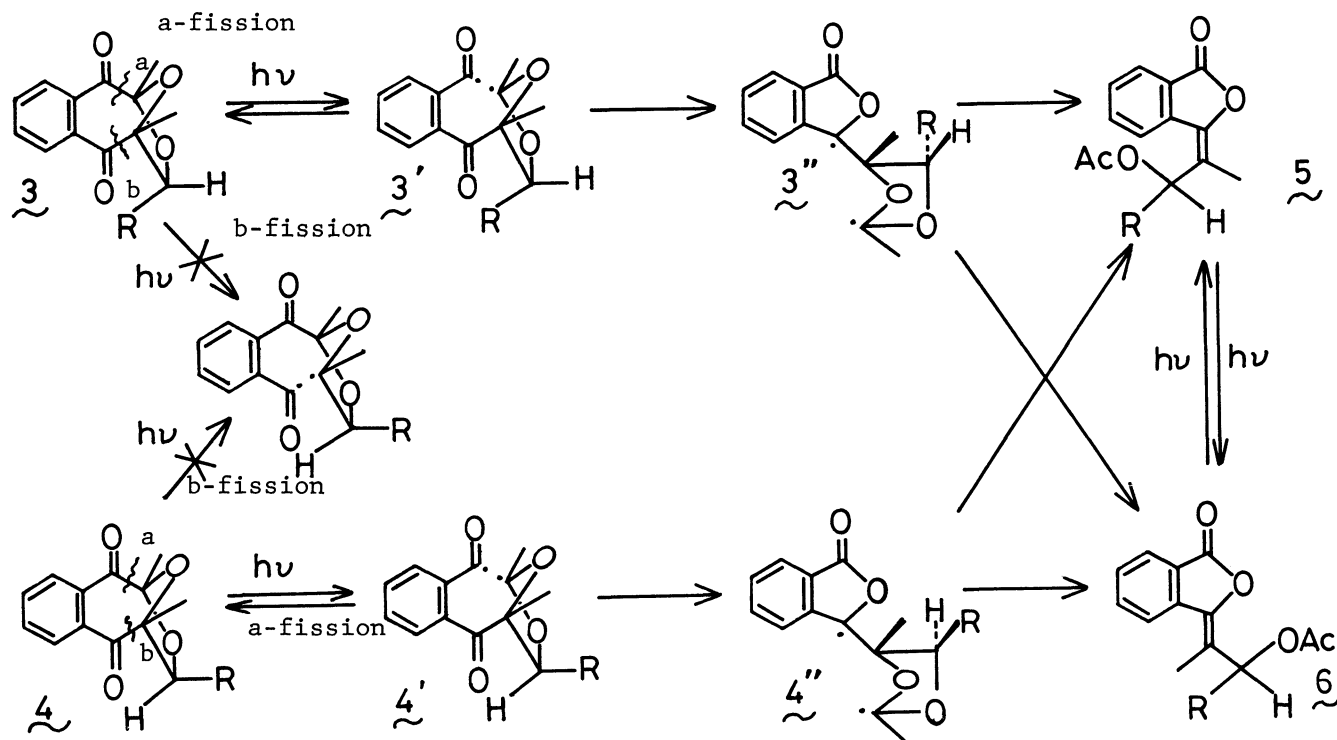
were made on the basis of their spectra (¹H NMR, IR, and mass) and elemental analyses.³⁾ Further, the alkylidene phthalide structure of 5a and 6a was evidenced by the ozonolysis of 5a and 6a, which gave phthalic anhydride 7 and 3-acetoxy-2-butanone (8). The allylic methine protons of 5a-h (H^a; designated in eq. 1) were deshielded relative to those of 6a-h without exception, because of the proximity of H^a to the benzene ring in 5a-h. The results are summarized in Table I.



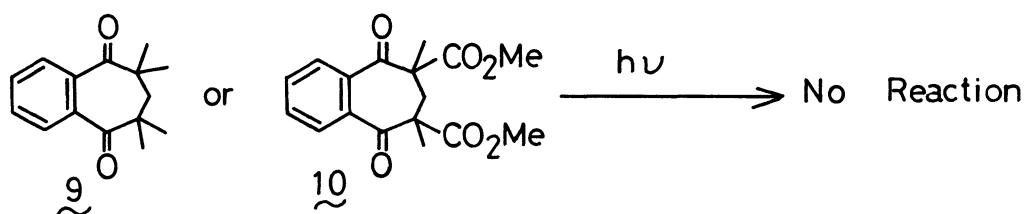
(E)-Alkylidene phthalide 5a photoisomerized to (Z)-isomer 6a and *vice versa*. After 15 hours, a photoequilibrated mixture of 5a and 6a (5a/6a=1.2) was formed. Even at low conversion of 3a and 4a (5%), under which conditions photo-interconversion between 5a and 6a was negligible, this product ratio (5a/6a) remained unchanged. Thus, 5a and 6a should be formed in the nearly same ratio from the photolysis of either 3a or 4a.



A plausible mechanism for the photoisomerization of 3 and/or 4 to 5 and 6 is shown in Scheme I. The photoisomerization could begin with α -cleavage, followed by lactonization to give 1,4-diradicals 3'' and 4'' that cleaved to give 5 and 6. The α -cleavage of alkyl phenyl ketones is known as a highly reversible process and the



Scheme I



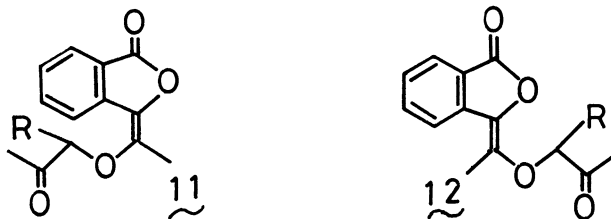
apparent inefficient quantum yields for the α -cleavage are explained in terms of the cage recombination of α -cleaved radicals⁴⁾. When the carbonyl group is present in enough proximity for the attack of the resulting acyl radical, some parts of **3a'** and **4a'** could undergo lactonization, leading to the formation of **3a''** and **4a''**, respectively. The pronounced preference of a-fission over b-fission, which is evident from the product distribution⁵⁾, is consistent with a stepwise radical rearrangement, since radical intermediate **3a'** and **4a'** may be fairly stabilized by the two etheral oxygen atoms adjacent to the radical center.

For comparison, 2,2,4,4-tetramethylbenzosuberone (**9**) and 2,4-dimethoxy-

carbonyl-2,4-dimethylbenzosuberan-1,5-dione (10)⁶⁾ were irradiated in benzene but no photoisomerization was observed even after 40 hours. Therefore, the radical stabilizing ability of the ethereal oxygen atoms situated β to carbonyl group or the strain arising from the bicyclic structure of 3 and 4 might be playing an important role in the facile photoisomerization of phthaloyl compounds to alkylidene phthalides.

References and Notes

- 1) L. F. Fieser and M. Fieser, "Advanced Organic Chemistry", Reinhold Publishing Corporation, New York, P. 796.
- 2) Bicyclo diketones 3a-h and 4a-h were obtained from the photo-induced cycloaddition of 2,3-dimethyl-2,3-epoxy-2,3-dihydro-1,4-naphthoquinone to aldehydes.
K. Maruyama and A. Osuka, Chem., Lett., 77 (1979).
- 3) Spectral data of 5a and 6a; 5a: IR (KBr) 3000, 1775(phthalide C=O), 1740(ester C=O), and 1238 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.50(d, J=7Hz, 3H), 2.09(s, 6H), 6.27(q, J=7Hz, 1H), and 7.44-8.02(m, 4H); MS (m/e) 246(M^+ , 25), 205(15), 204(75), 186(70), and 161(100).
6a: IR (KBr) 3000, 1775(phthalide C=O), 1740(ester C=O), and 1242 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.44(d, J=7Hz, 3H), 2.07(s, 3H), 6.14(q, J=7Hz, 1H), and 7.40-8.00 (m,4H); MS (m/e) 246(M^+ , 25), 204(64), 187(50), 186(75), and 161(100).
- 4) F. D. Lewis and J. G. Magyer, J. Am. Chem. Soc., 95, 5973 (1974).
- 5) Once being b-fission occurred, enol ketones 11 and 12 would be formed. Neither 11 nor 12 was found in the reaction.



- 6) G. L. Buchanan and J. K. Sutherland, J. Chem. Soc., 2620 (1956).

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